A VOCABULARY WORDS

Circle the word or words that complete the sentence. An example is done for you.

Example: Exposure to any kind of concrete / (radiation) can be dangerous.

- 1. Materials that have an unstable nucleus are called aluminum / radioactive.
- 2. Alpha particles are the most / least powerful type of radiation.
- 3. Gamma rays are the most / least powerful type of radiation.
- 4. Scientists wear special clothing when they work with <u>radioactive samples</u> / radioactive symbols.
- 5. Concrete / Paper can block beta particles.
- 6. Beta particles and gamma rays can pass through aluminum / paper.

B VOCABULARY IN CONTEXT

Choose a word or phrase from the box to complete the paragraph. An example is done for you.

radioactive sample	radioactive symbol	atom	beta particles
radiation	gamma rays	nucleus	alpha particles

Example: The center of an atom is called the _______

Sometimes an (1)	gives off particles and energ	y. These
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particles and energy are called (2) _____. The slowest type of

radiation is made up of (3) _____. (4) _____.

are smaller and faster. (5) ______ are the most powerful type of

radiation. Scientists wear special clothing when they handle a

- (6) ______. Radioactive materials are often marked with a
- (7) ______ to warn people of danger.

C DISCOVERY OF RADIOACTIVITY

Reading Strategy Cause and Effect

Read the Discovery of Radioactivity on page 184 in your book. Then answer the questions.

1. *Cause:* What was put in the drawer with the photographic plate?

Cause and Effect

- ✓ The cause tells what happened.
- ✓ The effect is the result of the cause.
- 2. Effect: What happened as a result? ______

D RADIOACTIVE DECAY

Reading Strategy Main Idea and Details

Read about Radioactive Decay on page 184 of your book. Then complete the chart with details that support the main idea. An example is done for you.

Main Idea and Details

- ✓ The main idea is the most important thing the paragraph says.
- ✓ Details support the main idea.

Main Idea: An atom decays when it gives off particles or energy.				
Detail An atom can release an alpha partícle.	Detail	Detail		

E SCIENCE SKILL Comparing Data

People are exposed to radiation every day. Some radiation comes from natural sources. Some radiation comes from human-made sources. The table at the top of the next page shows the main sources of a person's contact with radiation.

Compare Data

✓ Scientists often compare data. Placing data in a table makes the data easier to compare.

Source of Radiation	Natural or Human-Made	Percentage
radon (a gas found in the air)	natural	55%
inside human body	natural	11%
rocks and soil	natural	8%
outer space	natural	8%
consumer products	human-made	3%
medical	human-made	15%

- 1. What is the source of the most radiation?
- 2. What percentage of a person's contact with radiation comes from medical uses?
- **3.** What percentage of a person's contact with radiation comes from natural sources?

F USES OF RADIATION

Pairwork How People Use Radiation

Have you ever had an X-ray? Is there a smoke detector in your home? Work with a partner. Make a list of ways radiation helps people.

Apply Information ✓ Put information to use.

G WRITING Applying Information

What is the half-life of carbon-14? What does this mean? How do scientists use carbon-14? Write a paragraph.

GROUP WORK Model Half-Life

Question How can you make a model of half-life?

Procedure

- 1. Put 200 pennies tails up in a shoebox.
- 2. Put the cover on the box. Then, shake the box with one quick up-and-down motion.

Materials

- shoe box with cover
- 200 pennies
- **3.** Open the box. Remove all pennies that are heads up. These pennies model atoms that decayed.
- 4. Record the number of pennies removed. Then record the number of pennies left in the box. Use the data table below to organize your data.

Trial	Number of Pennies Removed	Number of Pennies Left in the Box
1		
2		
3		
4		
5		
6		
7		
8		
9		
10		

5. Do steps 2–4 nine more times.

Analysis

- 1. How many times did you have to shake the box to remove half the pennies?
- 2. Each shake represents one year. What is the half-life of the atoms?